# The Structure of Ethyl 5,6,7,8,9-Pentaacetoxy-2-butylamino-4-nitromethyl-d-glycero-L-gluco-2-nonene-3-carboxylate 

By R. Vega, A. López Castro and R. Márquez<br>Departamento de Optica y Sección de Física del Departamento de Investigaciones Físicas y Químicas del CSIC, Universidad de Sevilla, Spain

(Received 26 September 1980; accepted 18 November 1980)


#### Abstract

$\mathrm{C}_{27} \mathrm{H}_{42} \mathrm{~N}_{2} \mathrm{O}_{14}$ is orthorhombic, space group $P 2_{1} 2_{1} 2_{1}$, with $a=11.147$ (1), $b=37.350$ (2), $c=7.705$ (1) $\AA$, $V=3207.9 \AA^{3}, Z=4, D_{c}=1.28, D_{o}=1.29 \mathrm{Mg} \mathrm{m}^{-3}$, $F(000)=1420, \mu=0.094 \mathrm{~mm}^{-1}$. The final $R=0.069$ for 2135 observed reflections. The configuration around the chiral centres $C(8), C(7), C(6), C(5)$ and $C(4)$ of the sugar chain is RSSRS, that is D-glycero-L-gluco. The molecule has an intramolecular hydrogen bond between the N and O atoms of the butylamino and carbonyl groups respectively, showing a chelated structure. The packing of the molecules is governed by normal van der Waals contacts.


## Introduction

The title compound was an intermediate in a synthesis of (3-pentaacetoxypentyl)pyrrole, a substance which has aroused interest as a precursor of $C$-glycosylpyrroles. It was obtained by a Michael-type addition of 3-(butylamino)crotonic ester to D-galacto-penta-acetoxy-1-nitroheptene (Gómez-Sánchez, Mancera, Rosado \& Bellanato, 1980). In the assigned structure:

a new chiral centre is found (marked with an asterisk) and two diastereoisomers could result; however, only one product was isolated, and detected chromatographically. The configuration of this derivative could not be established from chemical and spectroscopic considerations. An X-ray analysis was undertaken to clarify
the problem and to determine conformational details of the molecule.

## Experimental

Crystals were in the form of colourless prisms. Preliminary cell dimensions and space-group information were obtained from oscillation and Weissenberg photographs taken with Mo Ka radiation. Accurate cell dimensions were obtained by least squares from the $\theta$ values of 50 reflections measured on a Philips PW 1100 automated four-circle diffractometer. Intensities were collected with graphite-monochromated Mo Ka radiation and the $\omega-2 \theta$ step scanning mode. Intensities were collected over the range $2<\theta<30^{\circ}$ to give 4997 independent reflections of which 2862 were considered as observed $[I>2 \sigma(I)]$.

## Structure determination

The structure was solved with MULTAN (Main, Lessinger, Woolfson, Germain \& Declercq, 1977). 330 reflections with $|E|>1.50$ (approximately seven reflections for each non-hydrogen atom in the asymmetric unit) were used and a choice of phasing was possible based on FOM's; the highest combined figure of merit derived with unit weights for the components was $2 \cdot 0$. An $E$ map computed with this phase set had distinguishable peaks for 40 of the 43 non-hydrogen atoms. The remaining atoms were located from a Fourier synthesis.

The structure was refined by full-matrix least squares. Scattering factors were taken from International Tables for X-ray Crystallography (1974). Refinement of the non-hydrogen atoms with isotropic temperature factors and unit weights gave $R=0 \cdot 12$. With anisotropic temperature factors and the weighting scheme $w=1 / \sigma^{2}(F), R$ was reduced to 0.071 . The positions of the H atoms were calculated. A difference synthesis revealed most of the H atoms near their calculated positions. Further refinement with non© 1981 International Union of Crystallography
hydrogen atoms treated anisotropically and H atoms isotropically produced convergence with $R=0.069$.

In the last cycle the H atoms were kept fixed with isotropic temperature factors equal to $4 \cdot 0 \AA^{2}$. The positional and isotropic thermal parameters for the non-hydrogen atoms are given in Table 1, those for the H atoms in Table 2.*

As can be seen from Table 1, the temperature factors for $C(24), C(33), O(41)$ and $O(42)$ atoms are higher than normal, possibly due to their location at the end of the chain.

> * Lists of structure factors and anisotropic thermal parameters have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 35905 (23 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

Table 1. Atomic coordinates $\left(\times 10^{4}\right)$ and isotropic temperature factors for non-hydrogen atoms

|  | $x$ | $y$ | $z$ | $B\left(\AA^{2}\right)$ |
| :---: | :---: | :---: | :---: | :---: |
| C(1) | 2770 (7) | 8113 (2) | 4005 (11) | 5.4 |
| C(2) | 3835 (6) | 8124 (2) | 2844 (9) | 3.8 |
| C(3) | 4958 (6) | 8270 (2) | 3347 (9) | $3 \cdot 6$ |
| C(4) | 5121 (6) | 8416 (2) | 5190 (8) | $3 \cdot 5$ |
| C(5) | 5663 (6) | 8793 (2) | 5236 (8) | $3 \cdot 1$ |
| C(6) | 4963 (5) | 9069 (2) | 4177 (8) | $3 \cdot 3$ |
| C(7) | 5381 (6) | 9449 (2) | 4576 (9) | $3 \cdot 5$ |
| C(8) | 4689 (6) | 9753 (2) | 3691 (8) | $3 \cdot 6$ |
| C (9) | 5267 (8) | 10102 (2) | 4131 (10) | 4.7 |
| N(21) | 3687 (6) | 7988 (2) | 1245 (8) | $4 \cdot 8$ |
| C(21) | 2629 (7) | 7791 (2) | 0616 (12) | $5 \cdot 3$ |
| C(22) | 1818 (7) | 8024 (2) | -0499 (11) | $5 \cdot 2$ |
| C(23) | 0756 (9) | 7805 (3) | -1203 (14) | $7 \cdot 6$ |
| $\mathrm{C}(24)$ | -0086 (10) | 8012 (3) | -2342 (15) | 8.1 |
| O(31) | 7005 (4) | 8338 (1) | 2936 (6) | 4.5 |
| $\mathrm{O}(32)$ | 5936 (4) | 8140 (1) | 0674 (6) | $5 \cdot 0$ |
| C(31) | 5941 (6) | 8234 (2) | 2201 (10) | $3 \cdot 8$ |
| C(32) | 8097 (7) | 8277 (3) | 1945 (13) | 7.1 |
| C(33) | 9086 (8) | 8396 (4) | 2936 (16) | $9 \cdot 8$ |
| $\mathrm{O}(41)$ | 4393 (7) | 7779 (2) | 7167 (11) | 8.1 |
| $\mathrm{O}(42)$ | 5893 (8) | 7561 (2) | 5776 (11) | 9.7 |
| $\mathrm{N}(41)$ | 5363 (8) | 7801 (2) | 6427 (10) | 5.9 |
| $\mathrm{C}(41)$ | 5920 (7) | 8169 (2) | 6330 (10) | 4.2 |
| $\mathrm{O}(51)$ | 5667 (4) | 8912 (1) | 7058 (5) | 3.4 |
| O(52) | 7659 (4) | 8839 (2) | 7194 (7) | 5.9 |
| C(51) | 6756 (6) | 8934 (2) | 7862 (10) | 4.5 |
| C(52) | 6624 (8) | 9071 (2) | 9669 (10) | $5 \cdot 5$ |
| O(61) | 3695 (4) | 9060 (1) | 4608 (6) | 3.8 |
| O(62) | 3225 (5) | 8962 (2) | 1780 (7) | 5.6 |
| C(61) | 2917 (6) | 9021 (2) | 3213 (11) | 4.9 |
| C(62) | 1643 (7) | 9084 (3) | 3853 (15) | 8.1 |
| $\mathrm{O}(71)$ | 6627 (4) | 9456 (1) | 4016 (6) | 3.5 |
| $\mathrm{O}(72)$ | 7176 (5) | 9718 (2) | 6551 (8) | $6 \cdot 1$ |
| C (71) | 7443 (6) | 9589 (2) | 5172 (11) | 4.6 |
| C(72) | 8684 (7) | 9550 (3) | 4456 (13) | $6 \cdot 6$ |
| $\mathrm{O}(81)$ | 4745 (4) | 9699 (1) | 1818 (5) | $3 \cdot 5$ |
| $\mathrm{O}(82)$ | 2761 (5) | 9811 (2) | 1535 (8) | $6 \cdot 2$ |
| C(81) | 3701 (6) | 9735 (2) | 0892 (10) | 4.1 |
| C(82) | 3921 (8) | 9673 (2) | -0994 (10) | 5.4 |
| O(91) | 4520 (5) | 10388 (1) | 3468 (6) | $5 \cdot 0$ |
| O(91) | 5284 (6) | 10746 (2) | 5485 (9) | 7.3 |
| C(91) | 4625 (9) | 10700 (2) | 4268 (12) | 5.9 |
| C(92) | 3846 (12) | 10986 (2) | 3516 (14) | $7 \cdot 8$ |

Table 2. Fractional atomic coordinates $\left(\times 10^{3}\right)$ for H atoms

|  | $x$ | $y$ | $z$ |  | $x$ | $y$ | $z$ |
| :--- | ---: | ---: | ---: | :--- | :---: | :---: | ---: |
| H1(C1) | 204 | 828 | 337 | H2(C32) | 808 | 840 | 063 |
| H2(C1) | 243 | 785 | 413 | H11(C33) | 901 | 868 | 327 |
| H3(C1) | 284 | 823 | 530 | H2(C33) | 914 | 825 | 419 |
| H(C4) | 426 | 841 | 589 | H3(C33) | 991 | 835 | 223 |
| H(C5) | 658 | 875 | 473 | H1(C41) | 610 | 828 | 760 |
| H(C6) | 513 | 899 | 281 | H2(C41) | 684 | 816 | 583 |
| H(C7) | 527 | 951 | 595 | H1(C52) | 622 | 933 | 963 |
| H(C8) | 376 | 974 | 418 | H2(C52) | 603 | 889 | 1041 |
| H1(C9) | 620 | 1012 | 361 | H31C52) | 747 | 909 | 1034 |
| H2(C9) | 550 | 1013 | 550 | H11(C62) | 141 | 888 | 486 |
| H(N21) | 441 | 803 | 031 | H2(C62) | 155 | 935 | 441 |
| H1(C21) | 288 | 754 | -008 | H3(C62) | 100 | 905 | 281 |
| H2(C21) | 213 | 767 | 166 | H1(C72) | 891 | 927 | 417 |
| H1(C22) | 153 | 826 | 018 | H2(C72) | 877 | 970 | 322 |
| H2(C22) | 232 | 815 | -155 | H3(C72) | 937 | 965 | 533 |
| H1(C23) | 028 | 767 | -012 | H1(C82) | 459 | 986 | -147 |
| H2(C23) | 106 | 756 | -188 | H2(C82) | 428 | 940 | -120 |
| H1(C24) | -046 | 825 | -155 | H3(C82) | 312 | 970 | -177 |
| H2(C24) | 037 | 813 | -342 | H1(C92) | 403 | 1102 | 217 |
| H3(C24) | -086 | 787 | -279 | H2(C92) | 289 | 1089 | 363 |
| H1(C32) | 821 | 799 | 149 | H3(C92) | 386 | 1124 | 416 |



## Results and discussion

The molecule consists of five acetoxy, a butylamino, an ethoxycarbonyl, and a nitromethyl group attached to a main nonene chain. Bond distances and angles not involving H atoms are shown in Fig. 1. The average e.s.d.'s in these bond lengths and angles are $0.009 \AA$ (maximum $0.015 \AA$ ) and $0.6^{\circ}$ (maximum $0.9^{\circ}$ ) respectively. Bond lengths and angles are quite close to those found in other compounds (Vloom, Hottentot, Overbeek \& Stam, 1979). The only significant differences are in angles which are more responsive to differences in packing and conformation.

The main chain has an approximately planar zigzag part comprising $C(4), C(5), C(6), C(7), C(8)$ and $C(9)$ with deviations $<0.12 \AA$ from their best plane and a second one, approximately planar too, involving C(1), $\mathrm{C}(2), \mathrm{C}(3)$ and $\mathrm{C}(4)$ with a largest deviation of $0.009 \AA$. The two chain sections are joined by $\mathrm{C}(3)-\mathrm{C}(4)$ (torsion angle $129.9^{\circ}$ ). The five acetoxy groups are planes with deviations $<0.016 \AA$. The butyl chain is planar and the N atom attached to it is at $0.07 \AA$. The nitromethyl group is also planar, maximum deviation from the best plane $0.002 \AA$. The ethoxycarbonyl group is not planar.

The acetoxy groups are markedly twisted out from the chain and the torsion angles are: $\mathrm{C}(9)-\mathrm{C}(8)-$ $\mathrm{O}(81)-\mathrm{C}(81)=-107.0(7), \mathrm{C}(8)-\mathrm{C}(7)-\mathrm{O}(71)-\mathrm{C}(71)$ $=104 \cdot 1(6), \mathrm{C}(7)-\mathrm{C}(6)-\mathrm{O}(61)-\mathrm{C}(61)=112 \cdot 2(6)$, $\mathrm{C}(6)-\mathrm{C}(5)-\mathrm{O}(51)-\mathrm{C}(51)=-127.4$ (6) and $\mathrm{C}(8)-$ $\mathrm{C}(9)-\mathrm{O}(91)-\mathrm{C}(91)=155 \cdot 1(6)^{\circ}$. The butylamino, ethoxycarbonyl and nitromethyl groups are slightly twisted from the chain; the torsion angles are: $\mathrm{C}(3)-\mathrm{C}(2)-\mathrm{N}(21)-\mathrm{C}(21)=172 \cdot 3$ (6), $\mathrm{C}(4)-\mathrm{C}(3)-$ $\mathrm{C}(31)-\mathrm{O}(31)=1 \cdot 0(9), \mathrm{C}(4)-\mathrm{C}(3)-\mathrm{C}(31)-\mathrm{O}(32)=$ -175.9 (7) and $\mathrm{C}(5)-\mathrm{C}(4)-\mathrm{C}(41)-\mathrm{N}(41)=175.8$ (6) ${ }^{\circ}$. On the other hand, the torsion angle $\mathrm{C}(1)-$ $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(4)=2.0(9)^{\circ}$.

The configuration around the chiral centres of the main chain can be deduced from Newman projections (Fig. 2). In the projection along $\mathrm{C}(2)=\mathrm{C}(3), \mathrm{C}(1)$ and $\mathrm{N}(21)$ are nearly eclipsed by $\mathrm{C}(4)$ and $\mathrm{C}(31)$ with $\mathrm{C}(4)$ synperiplanar to $\mathrm{C}(1)$ and $\mathrm{C}(4)$, antiperiplanar to $\mathrm{N}(21)$. With respect to $\mathrm{C}(3)-\mathrm{C}(4), \mathrm{C}(2)$ is eclipsed by $\mathrm{HC}(4)$ with $\mathrm{C}(5)$ synclinal to $\mathrm{C}(31)$ and $\mathrm{C}(5)$ anticlinal to $\mathrm{C}(2)$. These peculiarities are attributed to the double bond $\mathrm{C}(2)=\mathrm{C}(3)$ which necessarily restricts the side-chain conformation. In the $\mathrm{C}(4)-\mathrm{C}(5)$ bond, $C(6)$ and $C(3)$ assume synclinal conformations and $\mathrm{C}(6)$ is antiperiplanar to $\mathrm{C}(41)$. The conformation about $\mathrm{C}(5)-\mathrm{C}(6)$ is such that for the two acetoxy groups $O(51)$ and $O(61)$ are in synclinal orientation and $C(7), C(4)$ antiperiplanar. A similar situation is found around $C(7)-C(8)$, with $O(71)$ synclinal to $O(81)$ and $C(9)$ antiperiplanar to $C(6)$. Also the conformation viewed down $C(7)-C(6)$ has $O(61)$ antiperiplanar to $\mathrm{O}(71)$ and $\mathrm{C}(5)$ antiperiplanar to $\mathrm{C}(8)$. In the last


conformational situation observed $\mathrm{C}(7)$ adopts an antiperiplanar orientation to $\mathrm{O}(91)$, and $\mathrm{O}(81)$ is synclinal to $\mathrm{O}(91)$. Thus the configuration around the chiral centres $\mathrm{C}(8), \mathrm{C}(7), \mathrm{C}(6), \mathrm{C}(5)$ and $\mathrm{C}(4)$ of the sugar chain is RSSRS in accord with the D-glyceroL -gluco distribution.
The structure viewed down $\mathbf{c}$ is given in Fig. 3. The molecule has an intramolecular hydrogen bond between the N and O atoms of the butylamino and carbonyl groups respectively: $\mathrm{N}(21) \cdots \mathrm{O}(32)=2 \cdot 607$, $\mathrm{N}(21)-\mathrm{HN}(21)=1.094, \mathrm{HN}(21) \cdots \mathrm{O}(32)=1.768 \AA$, $\mathrm{HN}(21)-\mathrm{O}(32)-\mathrm{N}(21)=31^{\circ}$. This confirms the chelated structure previously observed by IR spectroscopy (Gómez-Sánchez et al., 1980). The packing of the molecules is governed by normal van der Waals contacts.
All calculations were carried out on a DCT 2000 terminal of the Centro de Calculo of this University, connected to a Univac 1108 computer, with XRAY (1970).


Fig. 3. The crystal structure.

The authors thank Professor A. Gómez-Sánchez for suggesting the study and supplying the crystal, and Professor A. Conde for helpful discussions, and the staff of Instituto 'Rocasolano' CSIC (Madrid), especially $\operatorname{Dr}$ F. Cano, for collecting diffractometer data.

The present work is part of a wider research project supported by the Government through the Comision Asesora de Investigación Cientifica y Técnica.

## References

Gómez-Sánchez, A., Mancera, M., Rosado, F. \& Bellanato, J. (1980). J. Chem. Soc. Perkin Trans. 1, pp. 1199-1205.
International Tables for X-ray Crystallography (1974). Vol. IV, pp. 72-78. Birmingham: Kynoch Press.
Main, P., Lessinger, L., Woolfson, M. M., Germain, G. \& Declerce, J. P. (1977). MULTAN 77. A System of Computer Programs for the Automatic Solution of Crystal Structures from X-ray Diffraction Data. Univs. of York, England, and Louvain-la-Neuve, Belgium.
Vloom, W.J., Hottentot, D., Overbeek, A. R. \& Stam, C. H. (1979). Acta Cryst. B34, 119-121.

XRAY (1970). Edited by J. M. Stewart, F. A. Kundell \& J. C. Baldwin. Computer Science Center, Univ. of Maryland, College Park, Maryland.

# The Structure of the Antibiotic Amicetin Consisting of Nucleobase, Disaccharide and Amino Acid Moieties 

By Janet L. Smith $\dagger$ and M. Sundaralingam $\ddagger$<br>Department of Biochemistry, College of Agricultural and Life Sciences, University of Wisconsin - Madison, Madison, Wisconsin 53706, USA

(Received 3 September 1980; accepted 25 November 1980)


#### Abstract

Amicetin $\quad\left(\mathrm{C}_{29} \mathrm{H}_{42} \mathrm{~N}_{6} \mathrm{O}_{9}\right)$, 1-[4-O-(4,6-dideoxy-4-dimethylamino- $\alpha$-D-glucopyranosyl)-2,3,6-trideoxy- $\beta$-D-erythro-hexopyranosyl]- $N^{4}$-[4-(2-methyl-L-serylamino)benzoyllcytosine, is one of a group of cytosine nucleoside antibiotics which inhibit peptide bond formation during protein biosynthesis. It contains a nucleobase, a disaccharide and an amino acid moiety. The compound crystallizes in the orthorhombic space group $P 2_{1} 2_{1} 2_{1}$ with $a=25.793(5), b=$ 11.897 (2), $c=10 \cdot 185$ (2) $\AA, Z=4 ; ~ D_{m}=1.293$, $D_{c}=1.315 \mathrm{Mg} \mathrm{m}^{-3}$ and $V_{c}=3125.4(1.7) \AA^{3}$. The crystal structure was solved by direct methods and refined to an $R$ value of $0 \cdot 06$. The molecule is in a highly extended conformation with an end-to-end

^[ $\dagger$ Present address: Laboratory for the Structure of Matter, Code 6030, Naval Research Laboratory, Washington, DC 20375, USA. $\ddagger$ To whom correspondence should be addressed. ]


distance of $25 \AA$. The terminal hydroxyl group $\mathrm{O}(17)-\mathrm{H}$ of the $\alpha$-methyl-L-serine moiety exhibits a twofold conformational disorder (occupancy factors of 0.62 and 0.38 ) and both hydroxyl positions participate in intermolecular hydrogen bonds. Concomitant with this disorder, the $C(17)$ atom attached to the hydroxyl group also appears to be disordered. The cytosine base is in the anti conformation $\lfloor\chi=$ $25.7(5)^{\circ}$ ] with respect to the trideoxyglucose ring. The amide group $\mathrm{C}(4)-\mathrm{N}(4)-\mathrm{C}(7)-\mathrm{O}(7)$ is markedly distorted from planarity $\left[\omega=-159 \cdot 2(7)^{\circ}\right]$. The disaccharide component possesses a $\mathrm{C}(1)$ axial$C(4)$ equatorial bridge, with the trans conformation for the bridge torsions $\mathrm{C}\left(2^{*}\right)-\mathrm{C}\left(1^{*}\right)-\mathrm{O}\left(1^{*}\right)-$ $\mathrm{C}\left(4^{\prime}\right)\left[-163 \cdot 8(5)^{\circ}\right]$ and $\mathrm{C}\left(1^{*}\right)-\mathrm{O}\left(1^{*}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ $\left[-162.8(5)^{\circ}\right]$, and a +gauche conformation for the bridge torsions $\mathrm{O}\left(5^{*}\right)-\mathrm{C}\left(1^{*}\right)-\mathrm{O}\left(1^{*}\right)-$ $\mathrm{C}\left(4^{\prime}\right)\left[72 \cdot 6(5)^{\circ}\right]$ and $\mathrm{C}\left(1^{*}\right)-\mathrm{O}\left(1^{*}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)$ [76.4 (5) ${ }^{\circ}$ ]. An unusual feature of the crystal structure is that only four of the eight available protons take part

